



# Decarbonylative Synthesis of Aryl Nitriles from Aromatic Esters and Organocyanides by a Nickel Catalyst

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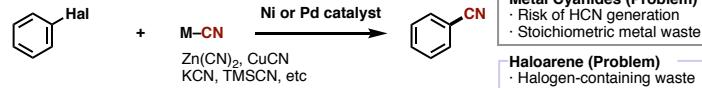
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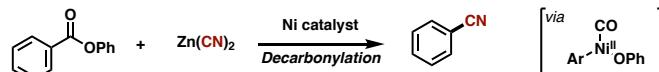


## Ni- and Pd-Catalyzed Cyanation using Metal Cyanides

### Cyanation of Haloarenes<sup>[1]</sup>

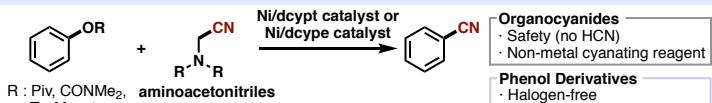


### Decarbonylative Cyanation of Phenyl Esters<sup>[2]</sup>

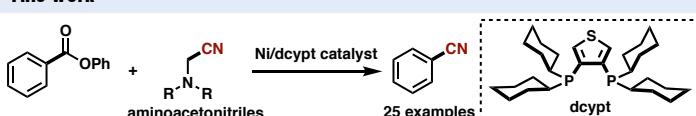


[1] (a) Ellis, G. P.; Romney-Alexander, T. M. *Chem. Rev.* 1987, 87, 779. (b) Anbarasan, P.; Schareina, T.; Beller, M. *Chem. Soc. Rev.* 2011, 40, 5049. (c) Wen, Q.; Jin, J.; Zhang, L.; Luo, Y.; Lu, P.; Wang, Y. *Tetrahedron Lett.* 2014, 55, 1271. (d) Yan, G.; Zhang, Y.; Wang, J. *Adv. Synth. Catal.* 2017, 359, 4068. [2] Chatupheeraphat, A.; Liao, H.-H.; Lee, S.-C.; Rueping, M. *Org. Lett.* 2017, 19, 4255.

## Our Previous Work<sup>[3]</sup>



## This Work<sup>[4]</sup>

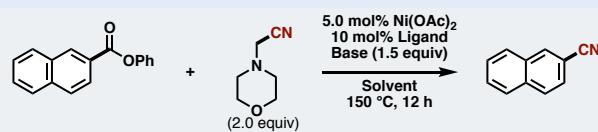


[3] Takise, R.; Itami, K.; Yamaguchi, J. *J. Org. Lett.* 2016, 18, 4428.

[4] Izumi, K.; Kurosawa, M. B.; Isshiki, R.; Muto, K.; Yamaguchi, J. submitted.

## Results

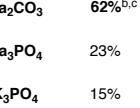
### Condition Screening



#### Ligand

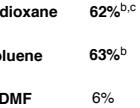


#### Base



with dcypt in 1,4-dioxane.

#### Solvent

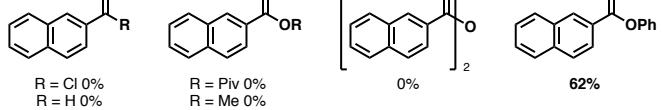


with Na<sub>3</sub>PO<sub>4</sub> in 1,4-dioxane. <sup>a</sup> NaOt-Bu (20 mol%) was added.

<sup>b</sup> with dcypt and Na<sub>2</sub>CO<sub>3</sub>.  
<sup>c</sup> Ni(OAc)<sub>2</sub>: 10 mol%, dcypt: 20 mol%, 150 °C.

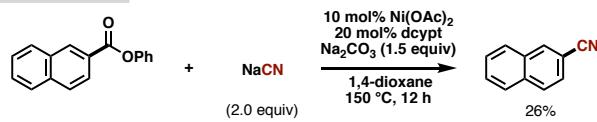
### Reactions Using Arylcarboxylic Acid Derivatives

#### Arylcarboxylic acid derivatives

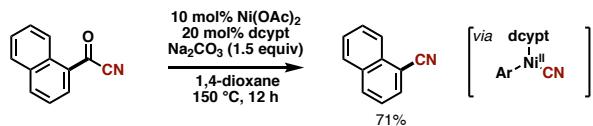


### Mechanistic Studies

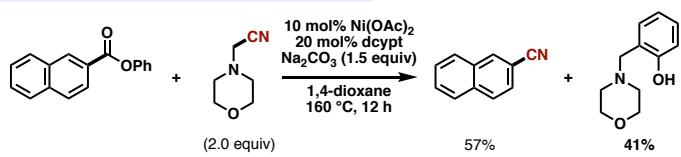
#### A. Reaction with NaCN



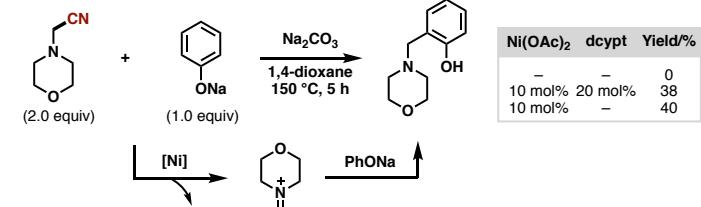
#### B. Reaction of Acylnitrile



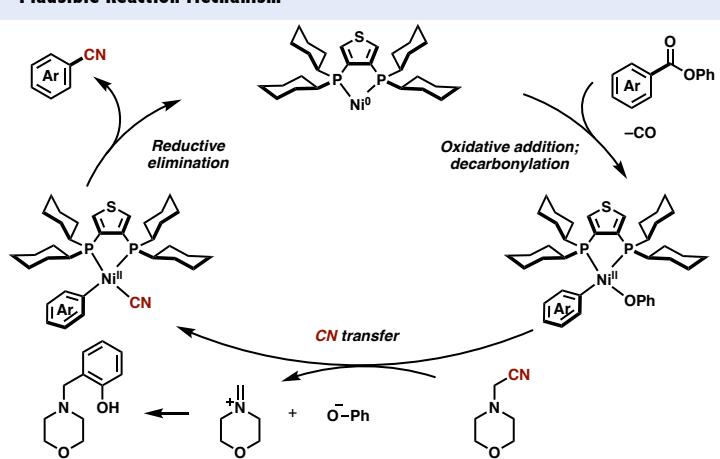
#### C. Aminoalkylphenol as a Coproduct of Aryl Nitrile



#### D. A Postulated Mechanism of Cyanide Generation

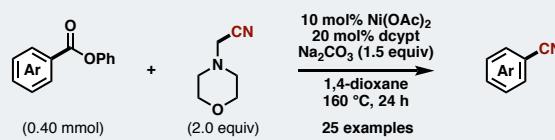


### Plausible Reaction Mechanism

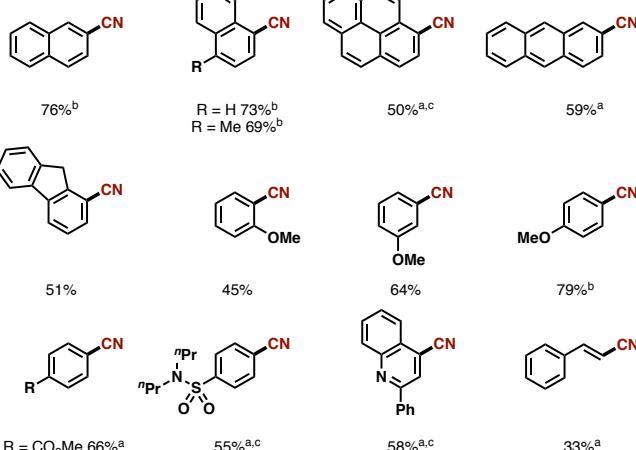


Ni(OAc)<sub>2</sub> (10 mol%), dcypt (20 mol%), Na<sub>2</sub>CO<sub>3</sub> in 1,4-dioxane.

### Substrate Scope



#### Selected examples (Arylnitriles)



<sup>a</sup> 12 h. <sup>b</sup> Toluene instead of 1,4-dioxane. <sup>c</sup> 150 °C.

R = CO<sub>2</sub>Me 66%<sup>a</sup>  
R = CN 68%<sup>a,c</sup>  
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